This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

SYNTHESIS AND NMR-ANALYSIS OF TRICYCLIC NUCLEOSIDES

Poul Nielsen^a; Michael Petersen^a; Jens Peter Jacobsen^a

^a Department of Chemistry, University of Southern Denmark, Odense M, Denmark

Online publication date: 31 March 2001

To cite this Article Nielsen, Poul , Petersen, Michael and Jacobsen, Jens Peter (2001) 'SYNTHESIS AND NMR-ANALYSIS OF TRICYCLIC NUCLEOSIDES', Nucleosides, Nucleotides and Nucleic Acids, 20: 4, 1309 - 1312

To link to this Article: DOI: 10.1081/NCN-100002543 URL: http://dx.doi.org/10.1081/NCN-100002543

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

SYNTHESIS AND NMR-ANALYSIS OF TRICYCLIC NUCLEOSIDES

Poul Nielsen,* Michael Petersen, and Jens Peter Jacobsen

Department of Chemistry, University of Southern Denmark, 5230 Odense M, Denmark

ABSTRACT

Two anomeric tricyclic nucleosides have been synthesised from diacetone-D-glucose using oxidation, stereoselective Grignard-addition of a vinyl-group, a stereoselective dihydroxylation followed by a tandem ring closing reaction, and finally a nucleobase coupling. The main β -configured product was examined and its configuration confirmed using NMR-spectroscopy in connection to *ab initio* calculations. The preferred conformation of this tricyclic nucleoside was described.

Conformationally restricted nucleosides have been intensively investigated as potential antiviral agents and in oligonucleotide analogues (1). Thus, bi- and tricyclic nucleosides have been constructed and further developed into nucleic acid analogues with very promising abilities in the recognition of complementary nucleic acid sequences (2–6). As an example, the bicyclic nucleoside 1 (Scheme 1) has been synthesised and incorporated into oligodeoxynucleotides (2). A fully modified sequence demonstrated moderately enhanced affinity towards complementary RNA. The furanose ring in 1 has been shown to prefer an O4'-endo conformation as demonstrated by an X-ray crystallographic study of 1 incorporated in a DNA dodecamer duplex (7) as well as by a corresponding NMR-study (8).

In order to improve the properties of 1, we decided to introduce further conformational restriction into this bicyclic structure by the synthesis of tricyclic nucleosides (Scheme 1). As a convenient and cheap starting material,

^{*}Corresponding author.

Scheme 1. a) i. CrO_3 , Ac_2O , Pyridine, CH_2Cl_2 , ii. VinylMgBr, Ether, THF, iii. 80% AcOH (75%); b) i. TrCl, Pyridine, ii. BnBr, NaH, DMF (63%); c) i. 20% HCl in MeOH, H_2O (76%), ii. MsCl, Pyridine (95%); d) i. OsO_4 , NMO, H_2O , Pyridine, t-BuOH, ii. NaH, DMF (48/42%); e) i. Thymine, BSA, TMS-Tf, CH_3CN (52%), ii. H_2 , $Pd(OH)_2$ -C, EtOH (73%). T = thymin-1-yl.

diacetone-D-glucose was chosen and stereoselectively converted to the 3'-C-vinyl compound 2 using a slightly changed literature method (9). Reprotection to give 3, acidic treatment and mesylation gave the separable anomers 4a and 4b. Dihydroxylation of 4a followed by a base-induced tandem ring closure afforded only one major product 5a. The exact configuration of 5a was not elucidated at this stage. However, MS-data and NMR strongly suggested the formation of a tricyclic structure. In a similar sequence, 4b afforded the other anomer 5b. A nucleobase coupling of either of the two anomers gave the same inseparable anomeric mixture of nucleosides which after deprotection were separated to give the two tricyclic nucleoside products 6 and 7 in a 1:3 ratio.

The exact configurations of **6** and **7** could only partly be confirmed by NOE-spectroscopy due to spectral overlap. Thus, an NOE-contact between H1' and H4' was only seen for the major product (**7**) confirming its β -configuration. However, the exact ${}^3J_{HH}$ coupling constants were measured for both compounds as shown for **7** in Table 1. Subsequently, *ab initio* calculations were performed for **7** as well as for three alternative tricyclic structures **A**, **B** and **C**, which from a synthetic point of view had to be considered (Fig. 1). The torsional angles were calculated as shown in Table 1. Furthermore, the Karplus relationships (10) between $J_{H1'H2'}$ and the pseudorotation angle P of the furanose ring (11) as well as between the other ${}^3J_{HH}$ coupling constants listed in Table 1 and the corresponding dihedral angles were derived. From these Karplus curves (not shown), the dihedral angles allowed by the experimental coupling constants could be found. In none of the calculations performed on **A**–**C**, a geometry was obtained in which all torsional angles could fit all the measured coupling constants simultaneously. Even though calculations were



Table 1. Exp. H-NMR Data for 7 and Calc. Data for the Theoretically Possible β-Nucleosides c

REPRINTS

δ/ppm	³ J _{HH} /Hz		7	A	В	С
H1′ 6.11	H1′H2′ 4.7	P	85°	39°	175°	85°
H2' 4.39		$\Phi_{ m max}$	37°	22°	23°	38°
H4' 4.04	H4'H5' 3.1	$\theta_{ m H4'H5'}$	55°	-26°	-42°	-23°
H5' 4.22	H5′H6′ 7.7	$\theta_{ ext{H5'H6'}}$	-23°	54°	84°	83°
H6' 4.02		$\theta_{ ext{H}5' ext{H}6''}$	-141°	-65°	−39°	− 39°
H6" 3.72	H5'H6" 8.0	$\theta_{ m H7'H8'}$	38°	44°	−55°	− 69°
H7′ 3.90	H7′H8′ 3.2	$ heta_{ ext{H7'H8''}}$	-85°	172°	62°	165°
H8' 4.01		γ	178°	92°	81°	95°
H8" 3.95	H7'H8" <1.5					

^aCD₃OD at 500 MHz.

performed in which some of the angles were constrained in allowed angles (not shown), other angles did not fit the experimental data. Thus, we conclude that 7 is the only possible structure fitting our experimental NMR-data and the configuration of 7 is hereby confirmed.

Since no change in the coupling constants of 7 was observed in the temperature range from -50 to 50° C, the nucleoside is believed to exist in only one conformation. This conformation is described by the torsional angles obtained from the unconstrained geometry optimisation (Table 1) and is shown in Figure 1. The furanose ring of this tricyclic nucleoside prefers the O4'-endo conformation as in the bicyclic nucleoside 1. This is an unusual and high-energy conformation in unmodified nucleosides (11). However, this conformation might be favourable for nucleic acid recognition as found for oligodeoxynucleotides containing 1 (2), its smaller analogue with a four-membered ring (3), or the 2'F-arabinodeoxynucleoside analogues (7,12). On the other hand, the C4'-C5' torsional angel γ of 7 is found to be in the +ap range which is probably unfavourable for Watson-Crick type duplex formation (6).

In conclusion, the alternative epimer A seems to be restricted in a more favourable conformation than 7 and the synthesis of this nucleoside as well as the

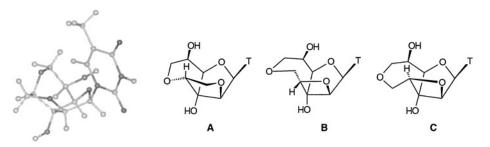


Figure 1. The determined conformation of 7, and the possible alternative structures.



^bPerformed at the 3–21G level.

^cFor definitions on P, Φ_{max} and γ see ref. 11.



NIELSEN, PETERSEN, AND JACOBSEN

incorporation of both epimers into oligonucleotide sequences will be performed in due course.

ACKNOWLEDGMENTS

The Danish Natural Science Research Council is thanked for financial support.

REFERENCES

1. Herdewijn, P. Liebigs Ann. 1996, 1337–1348.

1312

- 2. Nielsen, P.; Pfundheller, H. M.; Olsen, C. E.; Wengel, J. *J. Chem. Soc.*, *Perkin Trans. 1*, **1997**, 3423–3433.
- 3. Christensen, N. K.; Petersen, M.; Nielsen, P.; Jacobsen, J. P.; Olsen C. E. Wengel J. *J. Am. Chem. Soc.* **1998**, *120*, 5458–4562.
- 4. S. K. Singh, P. Nielsen, A. A. Koshkin and J. Wengel, Chem. Commun., 1998, 455–456.
- 5. Tarköy, M.; Leumann, C. Angew. Chem. Int. Ed. Engl. 1993, 32, 1432–1434.
- 6. Steffens, R.; Leumann, C. J. J. Am. Chem. Soc. 1997, 119, 11548–11549.
- 7. Minasov, G.; Teplova, M.; Nielsen, P.; Wengel, J.; Egli, M. *Biochemistry*, **2000**, *39*, 3525–3532.
- 8. Jørgensen, L. B.; Nielsen, P.; Wengel, J.; Jacobsen, J. P. *J. Biomol. Struc. Dyn.*, **2000**, in Press.
- Marco-Contelles, J.; Ruiz, P.; Martínez, L.; Martínez-Grau, A. Tetrahedron 1993, 49, 6669–6694.
- 10. Donders, L. A.; de Leeuw, F. A. A. M.; Altona, C. *Magn. Res. Chem.*, **1989**, 27, 556–563.
- 11. Saenger, W. Principles of Nucleic Acid Structure, Springer, New York, 1984.
- 12. Damha, M. J.; Wilds, C. J.; Noronha, A.; Brukner, I.; Borkow, G.; Arion, D.; Parniak, M. A. *J. Am. Chem. Soc.*, **1998**, *120*, 12976–12977.



Request Permission or Order Reprints Instantly!

Interested in copying and sharing this article? In most cases, U.S. Copyright Law requires that you get permission from the article's rightsholder before using copyrighted content.

All information and materials found in this article, including but not limited to text, trademarks, patents, logos, graphics and images (the "Materials"), are the copyrighted works and other forms of intellectual property of Marcel Dekker, Inc., or its licensors. All rights not expressly granted are reserved.

Get permission to lawfully reproduce and distribute the Materials or order reprints quickly and painlessly. Simply click on the "Request Permission/Reprints Here" link below and follow the instructions. Visit the U.S. Copyright Office for information on Fair Use limitations of U.S. copyright law. Please refer to The Association of American Publishers' (AAP) website for guidelines on Fair Use in the Classroom.

The Materials are for your personal use only and cannot be reformatted, reposted, resold or distributed by electronic means or otherwise without permission from Marcel Dekker, Inc. Marcel Dekker, Inc. grants you the limited right to display the Materials only on your personal computer or personal wireless device, and to copy and download single copies of such Materials provided that any copyright, trademark or other notice appearing on such Materials is also retained by, displayed, copied or downloaded as part of the Materials and is not removed or obscured, and provided you do not edit, modify, alter or enhance the Materials. Please refer to our Website User Agreement for more details.

Order now!

Reprints of this article can also be ordered at http://www.dekker.com/servlet/product/DOI/101081NCN100002543